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10. Explain in terms of the partial charge on hydrogen why NaOH is a base, HClO is a weak acid and HClO₄ is a strong acid. 11. Why is HCl a strong acid and HClO a weak acid? 12. Why are HCl and HClO₄ both strong acids? 13. For each of the reactions below, classify the reactants as an acid or a base

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and the products as
the conjugate acid or
...

**Acids and Bases 1
(Worksheet) -
Chemistry
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Brønsted-Lowry. The
Brønsted-Lowry
definition of acids and
bases liberates the
acid-base concept
from its limitation to
aqueous solutions, as
well as the
requirement that bases

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contain the hydroxyl group. A

Brønsted-Lowry acid is a hydrogen-containing species which is capable of acting as a proton (hydrogen ion) donor. A

Brønsted-Lowry base is a species which is capable of acting ...

Introduction to Acids and Bases (Worksheet) - Chemistry ...

Acids, Bases, Solutions,
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Titrations, Saturated,
Unsaturated,
Supersaturated This is a
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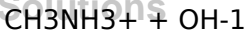
Kids

Acid and Base

Worksheet - Answers.

1) Using your knowledge of the Brønsted-Lowry theory of acids and bases, write equations for the following acid-base reactions and indicate each conjugate acid-base pair: a) $\text{HNO}_3 + \text{OH}^- \rightarrow \text{H}_2\text{O} + \text{NO}_3^-$. HNO_3 and NO_3^- make one pair OH^- and H_2O make the other. b) $\text{CH}_3\text{NH}_2 + \text{H}_2\text{O} \rightarrow$

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**Worksheet for Class
7 Science Chapter 5:
Acid Base and Salt**

Given acids or bases at
the same

concentration,

demonstrate

understanding of acid
and base strength by:

1. Relating the strength
of an acid or base to
the extent to which it
dissociates in water

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Worksheet
2. Identifying all of the molecules and ions that are present in a given acid or base solution. 3. Comparing the relative concentrations of molecules and ...

Acid-Base Solutions

- Acids | Bases |

Equilibrium - PhET ...

- a) Limewater is the common name for a diluted solution of calcium hydroxide
- c) Range 1-3 d) Strong

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Base is NaOH and
Weak Base is NH_4OH

e) A hydrated salt is copper sulphate crystals ($\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$). f) A universal indicator is a mixed indicator of organic chemicals which not only shows whether the given solution is acidic or basic, but also shows the approximate pH values by giving a wide ...

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**Acid base and Salts
Practice worksheet**

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5. An acid and a base neutralize each other

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and from a salt. True / False. 6. China rose indicator turns acidic solution to green. True / False. Please refer to attached file for Acids, Bases and Salts. CBSE Class 7 Science Worksheet - Acids, Bases and Salts

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an ionic compound that
can form from the
reaction of an acid with
a base. Salt. what is
the reaction between
an acid and a base.
Neutralization.

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Titrations, Saturated,
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Supersaturated This

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One of the ways pH
can be measured, is
with an acid-base

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indicator, such as universal indicator. An acid-base indicator is a substance that changes its colour depending on the pH of the solution that it is added to. The pH scale ranges between 1 and 14: Acids have pH values lower than 7; Bases have pH values higher than 7; and

**Summary | Acids,
Bases And The Ph
Value | Siyavula**

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