

## Activity Series Post Lab Answers Experiment 7

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### Activity Series Post Lab Answers

Create an activity series for the seven metals in this experiment by listing them from most reactive (at the top) to least reactive (at the bottom). 2. Answer the following questions based the activity series you just created. The word observed means what changes did you see (such as color changes, solid formation, etc.) Write the equations ...

### Metal Activity Series - Postlab Questions

The purpose of the lab was to find which metal is the most reactive and which metal is the least reactive. It was known before the experiment that the metals used in the experiment are placed in the activity series from most active to least active as follows: magnesium, aluminum, zinc, and copper. The hypotheses formed were that zinc nitrate would react with aluminum and magnesium; aluminum nitrate would react with magnesium; copper nitrate would react with zinc, magnesium, and aluminum; and ...

### Activity Series Lab Answers | SchoolWorkHelper

Question: Laboratory 11 Post-Lab: Halogen Activity Series Questions 1. For The First Ten Reactions, Write A Balanced Reaction For Each Tube In Which A Reaction Occurred. If There Was No Reaction, Write NR.

### Solved: Laboratory 11 Post-Lab: Halogen Activity Series Qu ...

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### Activity Series Lab Answer Key PDF Download - OsvaldAjeet

Question: EXPERIMENT 6: RELATIVE REACTIVITIES OF METALS AND THE ACTIVITY SERIES Name: Instructor: Post-Lab Instructor When You Have Completed The Report.) PART A & B: REACTION WITH WATER Date: Section/Group: Report (Use The In-lab Observations To Complete The Laboratory Report. Turn In To Your Als That Reacted With Water In Order Of Decreasing Reactivity (most ...

### Solved: EXPERIMENT 6: RELATIVE REACTIVITIES OF METALS AND ...

Answers to questions in complete sentences . REVISED 12/2003 Activity Series Lab - Observations for Part 1 KNO 3 Mg(NO 3) 2 Zn(NO 3) 2 CuSO 4 AgNO 3 Distilled H 2O Copper Iron Magnesium Tin Zinc . REVISED 12/2003 Activity Series Lab - Observations for Part 2 Reaction with HCl Copper

### ACTIVITY SERIES LAB - Auburn School District

Calcium reacted with every metal ions solution it was placed in. This was a key observation for calcium to be placed at the top of the activity series, because it was able to react with magnesium, zinc, iron, tin, and copper ions. This activity series can be used to predict if a reaction will occur between any metals on this list. Post-lab questions:

### Chemistry Lab Report (The Activity Series) - Sarah Jackson

Some of the worksheets for this concept are Work lab equipment, Lab equipment, Lab safety work, Lab equipment activity, Lab safety work answers, Lab safety work, Safe science lab safety awareness, Section 2. Once you find your worksheet, click on pop-out icon or print icon to worksheet to print or download. Worksheet will open in a new window.

### Lab Equipment Answer Sheet Worksheets - Learny Kids

QUESTIONS TO ANSWER IN ANALYSIS AND INTERPRETATION. Write balanced and chemical equations of all the reactions you tried. Which of the metals you tests were able to replace the hydrogen in hydrochloric acid (HCl)? Create an activity series of the 4 metals you tested in your lab. The list should be from the most reactive metal to least reactive ...

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A ranking according to reactivity is called an activity series. For example, an activity series containing the elements calcium, gold, and iron would put the reactive calcium at the top, iron in the middle, and the unreactive gold at the bottom.

### AP Chemistry Lab 3 1 Activity Series of Metals and Nonmetals

The hypothesis has been proven after the experiment. The activity series were correctly found for the metals Magnesium, Zinc, Lead, and Cupper and, the halogens Chlorine, Bromine, and Iodine by performing a series of reactions. This lab has almost a hundred percent accuracy since all the data found had matched the actual activity series.

### An Activity Series - WordPress.com

Activity Series The purpose of this lab was to observe the reactions of metals and the importance of the activity series. Grade and Comments Darbi Jeremy Colleen Class KNO3 Mg(NO3)2 Zn(NO3)2 CuSO4 AgNO3 Distilled water Copper Iron Magnesium Tin Zinc Spot plate scoopula Students

### Activity Series Lab Report by Darbi Cook - Prezi

The metal activity series indicates the reactivity of metals and how strongly they are reduced. The series begins with lithium, potassium, calcium, and sodium.

### Metal Activity Series - Chemistry | Socratic

A solid different than the chromium seems to appear. When a piece of platinum is then placed in a solution containing tin ions, no reaction is observed. Order these metals as they would appear in an activity series. (You can check your answer by looking at the example series. from the background section.)

### Metal Activity Series - Prelab Questions

Overview:The students will develop their own activity series of metals based on lab results. Qualitative observations will be used. Tie-ins include single-replacement reactions (SRR), oxidation- reduction, practical applications such as galvanization, and predicting reactions.

### Developing the Activity Series - ASM International

An Activity Series AP Chemistry Laboratory #20 Catalog No. AP5914 Publication No. 10536A I n t r o d u c t i o n In this e x p e r i m e n t , a series of metals and a series of nonmetal halogens are studied to find their r e l a t i v e r e a c t i v i t i e s . The r e a c t i v i t y of the metals is determined by combining the metals with a complemen-

### An Activity Series

From this lab one can conclude that the activity series for the metals, from most active to least, is Magnesium (Mg), Zinc (Zn), Lead (Pb), Copper (Cu), Silver (Ag). The halogen activity series is Chlorine (Cl), Bromine (Br), Iodine (I). No sources of error. This experiment was qualitative, not quantitative.

### An Activity Series Lab by av s on Prezi Next

In lab you are asked to perform an experiment that requires 100.00 mL of a 0.350 M potassium chloride solution. All you have in the lab though is a 0.750 M potassium chloride solution. ... Get the right answer, fast. Ask a question for free Get a free answer to a quick problem. Most questions answered within 4 hours. OR. Find an Online Tutor ...

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