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Have you ever been injured and been given an instant cold pack or hot pack to place on the injury? These packs usually function by mixing water and a salt. Squeezing the pack breaks an inner packet, releasing

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This video is about the AP Chemistry Lab Experiment #7: The Stoichiometry of a Chemical Reaction.

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Activity Background

Have you ever spilled bleach on a piece of clothing, raced to the sink and rinsed it off, yet still had a spot of discoloration?

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moles Therefore, moles
= molarity (M) x
volume (L). Molarity =

Step 2: Convert moles

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of solute to mass of
solute. grams of solute

— mass of solute Moles

of solute x 1 mole

solute In this

experiment, the total

volume of reactants is

constant; however, the

volume of each

reactant varies from

one mixture to

another.

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Equilibrium equation shows when the rate of reactants equal to the rate of products. When temperature or concentration are changed, the stress is placed on either of the reactant or products side. To relieve the stress, the equilibrium shifts left or right. The rate on the either side

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occur equally at the same time,...

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Laboratory -

Experiment #3 - Types
of Chemical Reactions.

In this video you will
learn about five
different types of

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chemical reactions that

are: 1- Single ...

Answers

**Lab Experiment #3:
Types of Chemical
Reactions.**

Subject/Grade Level:
chemistry (preAP/AP)

TEKS Addressed: (1)
Scientific processes.

The student, for at
least 40% of
instructional time,
conducts laboratory
and field investigations
using safe,
environmentally

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appropriate, and
ethical practices. The

student is expected to:

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