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reactants required or product formed in a nonchemical process. It teaches you how to interpret chemical equations in terms of interacting moles, representative particles, masses, and gas volume at STP.

## **SECTION 12.1 THE ARITHMETIC OF EQUATIONS**

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#### **Chapter 12 4 Answers**

Stoichiometry 353 12.1 FOCUS

Objectives 12.1.1 Explain how balanced equations apply to both chemistry and everyday situations. 12.1.2 Interpret balanced chemical equations in terms of moles, representative particles, mass,



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and gas volume at STP. 12.1.3 Identify the quantities that are always conserved in chemical reactions. Guide for Reading

## **12.1 The Arithmetic of Equations 12**

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