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Chemistry Chapter 4 Study Guide

Chapter 4 - Chemical Compounds 43 To get a review of the most important topics in the chapter, fill in the blanks in the Key Ideas section. Work all of the selected problems at the end of the chapter, and check your answers with the solutions provided in this chapter of the study guide. Ask for help if you need it. Web Resources

Chapter 4 Chemical Compounds - An Introduction to Chemistry

Chemistry Study Guide Chapter 4 - Key 1. Define the following terms: (a) atom, (b) atomic number, (c) mass number, (d) isotopes (give 2 definitions), (e) atomic mass unit (amu), (f) nucleus, (g) spectroscopy, (h) ground state, (i) excited state, (j) atomic mass, (k) quantum leap, (l) photon.

Chemistry Study Guide Chapter 4 - Tri-Valley Local Schools

©2016 American Chemical Society Middle School Chemistry -www.middleschoolchemistry.com 319 Chapter 4, Lesson 4: Energy Levels, Electrons, and Covalent Bonding Key Concepts • ⁺e electrons on the outermost energy level of the atom are called valence electrons.

Chapter 4, Lesson 4: Energy Levels, Electrons, and ...

Chemistry Study Guide Chapter 4 - Key 1. Define the following terms: (a) atom, (b) atomic number, (c) mass number, (d) isotopes (give 2 definitions), (e) atomic mass unit (amu), (f) nucleus, (g) spectroscopy, (h) ground state, (i) excited state, (j) atomic mass, (k) quantum leap, (l) photon.

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Chapter 4 - An Introduction to Chemical Reactions 41 Exercises Key Exercise 4.1 - Balancing Equations: Balance the following chemical equations. (Obj 4) a. P4(s) + 6Cl2(g) 4PCl3(l) b. 3PbO(s) + 2NH3(g) 3Pb(s) + N2(g) + 3H2O(l) c. P4O10(s) + 6H2O(l) 4H3PO4(aq) d. 3Mn(s) + 2CrCl3(aq) 3MnCl2(aq) + 2Cr(s) e.

Chapter 4 An Introduction to Chemical Reactions

Democritus Reasoned that atoms were indivisible and indestructible. Atom The smallest particle of an element that retains its identity in a chemical reaction. Dalton Transformed Democritus ideas on atoms into a scientific theory. Dalton's Atomic Theory 1. All elements are composed of tiny indivisible particles Dalton's atomic theory 2. Atoms of the same element are ...

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42 Study Guide for An Introduction to Chemistry Section Goals and Introductions Section 4.1 Energy Goals To introduce the terms energy, kinetic energy, and potential energy. To introduce the Law of Conservation of Energy. To describe the relationships between stability, capacity to do work, and potential energy. To explain why breaking chemical bonds requires energy and why the formation of

Chapter 4 Modern Atomic Theory - An Introduction to Chemistry

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CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following