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is the\_\_ absorption of water into their solid structures. Elements tend to gain or lose electrons in order to acquire the electron configuration of a\_\_ noble gas. Ions that are formed from more than one atom are called\_\_

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chemical formula subscript of each element reveals number of atoms contained in a single molecule of a certain compound  
monoatomic ions ions formed from a single atom; remember to change nonmetals to -ide ending  
oxyanions polyatomic ions that contain oxygen  
binary ionic compounds compounds of 2 elements between metal (positive, s-block and p-block) and nonmetal [...]

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monoatomic ion. binary compounds.  
nomenclature. oxyanions. an ion formed

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from a single atom. (remember to change nonmetal.... compounds composed of two different elements; (compounds of 2.... naming system. polyatomic ions that contain oxygen.

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO<sub>3</sub> a. copper(II) carbonate Na<sub>2</sub>SO<sub>3</sub> b. sodium sulfite (NH<sub>4</sub>)<sub>3</sub>PO<sub>4</sub> c. ammonium phosphate SnS<sub>2</sub> d. tin(IV) sulfide HNO<sub>2</sub> e. nitrous acid 2.

## **7 Chemical Formulas and Chemical Compounds**

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron(III)

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sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom. (c) the charge on each Fe ion.

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Modern Chemistry 35 Chapter Test Name Class Date Chapter Test B, continued Write the orbital notation for the following elements in the space provided Modern chemistry chapter 7

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test answer key. 37. lithium, atomic number 3 38. carbon, atomic number 6 39 Modern chemistry chapter 7 test answer key. neon, atomic number 10 PART VI Write the answers to the questions on the line to the left, and ...

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answer about a material called  
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Name Class Date Chapter Test B,  
continued 18. When a strong acid is  
titrated with a weak base, the pH of the  
solution at the equivalence point is than  
7. 19. A is a highly purified solid used to  
check the concentration of a standard  
solution. 20. A 1 M solution of NaOH will  
have a pH that is

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Modern Chemistry • CHAPTER 7  
HOMEWORK 7-6 SKILL BUILDER  
Determine the oxidation numbers of

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these compounds and ions.  $N_2$   $N_3$ - $N_2O_3$   $KNO_3$   $NH_4NO_3$   $K_2Cr_2O_7$   $CO_3^{2-}$   $NH_4^+$  ... Circle the letter of the best answer. 1. Which is the Stock system name of  $SO_2$ ? a. sulfur dioxide b. sulfur trioxide c. sulfur(IV) oxide d. sulfur(II) oxide

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