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Brønsted-Lowry Acids & Bases Identify each species in the following equation as either the Brønsted-Lowry acid, the Brønsted-Lowry base, the conjugate acid, or the conjugate base. Identify the conjugate acid-base pairs in the reaction.  $\text{H}_2\text{SO}_4(\text{aq}) + \text{HPO}_4^{2-}(\text{aq}) \rightleftharpoons \text{HSO}_4^-(\text{aq}) + \text{H}_2\text{PO}_4^-(\text{aq})$  Strong vs. Weak Acids and Bases Strong ...

## **Chapter 15: Acids and Bases Acids and Bases**

Identify the conjugate acid-base pairs in this equilibrium.  $\text{NH}_2^- + \text{H}_2\text{O} \rightleftharpoons \text{NH}_3 + \text{OH}^-$  Answer:  $\text{H}_2\text{O}$  (acid) and  $\text{OH}^-$  (base);

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NH<sub>2</sub> – (base) and NH<sub>3</sub> (acid) Chemistry Is Everywhere:  
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Whether maintaining water quality in a fish tank or suffering from acid indigestion, we encounter the effects of acid-base chemistry in many common situations. This module explores acids and bases and describes how our understanding of acid-base chemistry has been refined over the centuries. The module

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explains how the pH scale indicates acidity. A table of pH values demonstrates how familiar ...

## **Acids and Bases I | Chemistry | Visionlearning**

Acids, Bases, and Salts 39,067 Many acids and bases occur naturally in nature, such as citric acid in fruits like orange, lemon etc, tartaric acid in tamarind, malic acid in apples and lactic acid in milk and milk products, hydrochloric acid in gastric juices. Similarly, many bases are found such as lime water.

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Salts and Acids and Bases Pass the Salt, Acid, and Base Please. In this section we'll talk about the dirty little secret that some molecules disguised as innocuous salts can have acidic or basic tendencies. Here, a salt will refer to a molecule that dissociates entirely when added to an aqueous solution, which in our case will be water.

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