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53. ChemQuest 17. Name: _____ Date:
_____ Hour: _____ Information: Separating

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Charges. Examine Figure 1 below where there are three pairs of metal spheres that have different amounts of charge on them. The spheres in diagram C are closer than the others because they have the strongest attraction.

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153 and 154: 153 11. Find the pH of a 0.85 M sol. Page 155 and 156: 155 Information: Mixing Strong Acid. Page 157 and 158: 157 ChemQuest 52 Name: _____ Page 159 and 160: 159 8. Determine whether each of th. Page 161 and 162: 161 ChemQuest 53 Name: _____ Page 163 and 164: 163 7. In the above examples, Examp

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49 ChemQuest 16 Name: _____ Page 51
and 52: 51 10. In a silicon atom, the
force. Page 53 and 54: 53 ChemQuest
17 Name: _____ Page 55 and 56: 55
Information: Trend in Ionization. Page 57
and 58: 57 3. Given the above diagrams,
how. Page 59 and 60: 59 ChemQuest 19

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This video is designed to help you complete the molarity chemquest found in Unit 11.

Molarity

PLEASE NOTE: If you have a question about these answers, it is your responsibility to come to office hours or ask during class work time. UNIT 12 -

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HW Practice Keys - ChemActivity 51: Cell Voltage - ChemActivity 50:
Electrochemical Cell UNIT 11 - HW
Practice Keys - ChemQuest 55: Free Energy - ChemQuest 54: 2nd Law of Thermodynamics

**HW Keys - Roosevelt High School AP
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118 ChemQuest 39 Name: _____ Date:
_____ Hour: _____ Information: Molarity
Concentration is a term that describes
the amount of solute that is dissolved in
a solution. Concentrated solutions
contain a lot of dissolved solute, but
dilute solutions contain only a little.

118 ChemQuest 39 Name:

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Find the mole fraction of water (X_{water}) for the solution described in question 12. $\frac{4.722 \text{ mol H}_2\text{O}}{4.722 \text{ mol H}_2\text{O} + 0.2436 \text{ mol NaCl}} = 0.9512$

Prove that $X_{\text{solute}} + X_{\text{solvent}} = 1$. The answer from question 12 + answer to question 13 = 1.15. In a certain salt water solution, the mole fraction of salt is 0.18.

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ChemQuest 39-41Answers - Yumpu

ChemQuest 5 Name: _____ Date: _____

Hour: _____ Information: Heat and

Temperature When a substance absorbs heat energy, the temperature of the substance increases. There are a number of factors (such as mass of the substance) that affect how much the

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temperature of a substance changes.
Critical Thinking Questions 1.

Critical Thinking Questions

An element exists as 4 different isotopes. 4.35% have a mass of 49.9461, 83.79% have a mass of 51.9405, 9.50% have a mass of 52.9407 amu, and 2.36% have a mass of 53.9389

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amu. a. What is the average atomic mass for this element? b. What is the identity of this element?

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ChemQuest 50 Diffusion 173

ChemQuest 51 Gas Stoichiometry ...

15. Given your answer to question 14 and the reaction you wrote in question 13, calculate the molarity of the nitrate ions (NO_3^-) in the solution. Title:

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