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Borgnakke and Sonntag 2.5 An electric dip heater is put into a cup of water and heats it from 20°C to 80°C. Show the energy flow(s) and storage and explain what changes. Solution: Electric power is converted in the heater element (an electric resistor) so it becomes hot and gives energy by heat transfer to the water.

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Borgnakke and Sonntag 1.d Density of liquid water is $\rho = 1008 - T/2$ [kg/m³] with T in °C. If the temperature increases, what happens to the density and specific volume? Solution: The density is seen to decrease as the temperature increases. $\Delta\rho = -\Delta T/2$ Since the specific volume is the inverse of the density $v = 1/\rho$ it will increase.

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Solution: We see hot high pressure steam flowing in at state 1 from the steam drum through a flow control (not shown). The steam leaves at a lower pressure to the condenser (heat exchanger) at state 2. A rotating shaft gives a rate of energy (power) to the electric generator set.

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