

Moles Worksheet Answer Key

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Moles Worksheet Answer Key

Mole Calculation Worksheet - Answer Key 1) How many moles are in 15 grams of lithium? 0.46 moles 2) How many grams are in 2.4 moles of sulfur? 77.0 grams 3) How many moles are in 22 grams of argon? 0.55 moles 4) How many grams are in 88.1 moles of magnesium? 2141 grams 5) How many moles are in 2.3 grams of phosphorus? 0.074 moles 6)

[PDF] Moles, Molecules, and Grams Worksheet Answer Key ...

Mole Calculation Worksheet - Answer Key What are the molecular weights of the following compounds? 1) NaOH $22.99 + 16.00 + 1.01 = 40.00$ grams/mol 2) H_3PO_4 $3(1.01) + 30.97 + 4(16.00) = 98.00$ grams 3) H_2O $2(1.01) + 16.00 = 18.02$ grams 4) Mn_2Se_7 $2(54.94) + 7(78.96) = 662.60$ grams 5) $MgCl_2 =$

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$24.31 + 2(35.45) = 95.21$ grams 6) $(\text{NH}_4)_2\text{SO}_4$

Mole Calculation Worksheet - Brookside High School

Moles, 'Molecules,' and 'Grams' Worksheet - 'Answer' Key

1) How many moles are there in 24.0 grams of FeF_3 ? .213 moles
2) How many moles are there in 458 grams of Na_2SO_4 ? 3.22 moles
3) How many grams are there in 2.30×10^{24} atoms of silver? 412 grams
4) How many grams are there in 7.40 moles of AgNO_3 ? 1260 grams (note: 3 significant figures)

Moles, Molecules, and Grams Worksheet and Key

Mole Calculation Worksheet - Answer Key

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Grams/Moles Calculations - Answer Key

Mole Calculation Worksheet - Answer Key 1) How many moles are in 15 grams of lithium? 0.46 moles 2) How many grams are in 2.4 moles of sulfur? 77.0 grams 3) How many moles are in 22 grams of argon? 0.55 moles 4) How many grams are in 88.1 moles of magnesium? 2141 grams 5) How many moles are in 2.3 grams of phosphorus? 0.074 moles

Mole Calculation Worksheet

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1 mole = molar mass (could be atomic mass from periodic table or molecular mass) 1 mole = 22.4 L of a gas at STP (You do not need to worry about this yet) Each definition can be written as a set of two conversion factors. They are: 1 mole = molar mass(g) can be written as $\frac{1 \text{ mole}}{\text{molar mass (g)}}$ OR $\frac{\text{molar mass (g)}}{1 \text{ mole}}$

Mole to Grams, Grams to Moles Conversions Worksheet

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Moles Worksheet (Solutions) 1) Define "mole". 6.02×10^{23} of anything, usually atoms or molecules. 2) How many moles are present in 34 grams of $\text{Cu}(\text{OH})_2$? 0.35 moles 3) How many moles are present in 2.45×10^{23} molecules of CH_4 ? 0.41 moles 4) How many grams are there in 3.4×10^{24} molecules of NH_3 ?

Moles Worksheet - Awesome Science Teacher Resources

1 mole = 22.4 L of a gas at STP (You do not need to worry about this yet) Each definition can be written as a set of two conversion factors. They are: 1 mole = molar mass (g) can be written as $\frac{1 \text{ mole}}{\text{molar mass (g)}}$ OR $\frac{\text{molar mass (g)}}{1 \text{ mole}}$.

Moles And Mass Worksheets - Kiddy Math

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Mole To Grams Grams To Moles Conversions Worksheet Answer Key

Determine the number of moles in each of the following. 20 A. 3.25×10 atoms Pb mol Ibu-cosz (B. 4.96×10^{24} molecules glucose C. 0.02×10^{23} 6.2 (o 23 C. 1.56×10 formula units NaOH 25 D. 1.25×10 cu Ions Part 3. Make the following conversions. A. 1.51×10^{15} atoms Si to mol Si C. oZKI B. 4.25×10^2 mol N₀₂ to molecules N₀₂ Imo (—

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