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## **Chemistry (12th Edition) Chapter 16 - Solutions - 16.2 ...**

Prentice Hall © 2005 General Chemistry 4th edition, Hill, Petrucci, McCreary, Perry Chapter Sixteen 16 • We generally consider precipitation to be “complete” if about 99.9% of the target ion is precipitated (0.1% or less left in solution). • Three conditions generally favor completeness of precipitation: 1. A very small value of  $K_{sp}$ . (few ions in solution)

## **Chapter Sixteen - University of Houston**

Chapter 16 Solutions167 SECTION 16.1 PROPERTIES OF SOLUTIONS (pages 471–477) This section identifies the factors that affect the solubility of a substance and determine the rate at which a solute dissolves. Solution Formation (pages 471–472)

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